

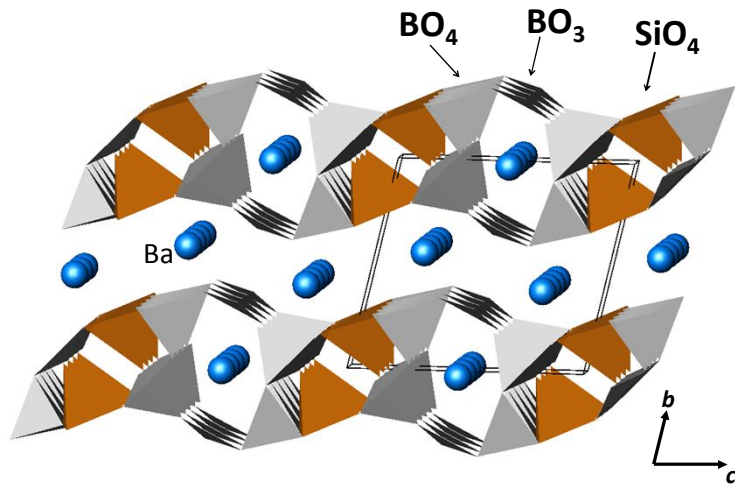
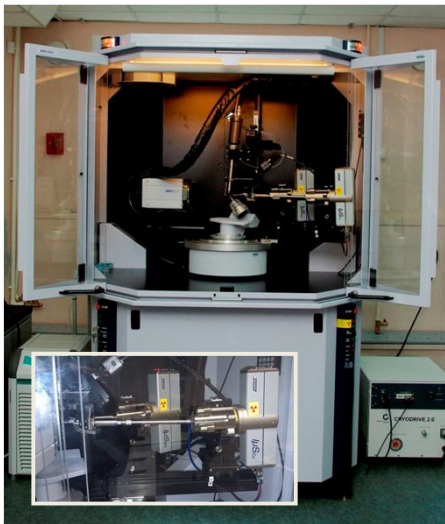


# Synthesis and crystal structure of $Ba_3B_6Si_2O_{16}$

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Single crystals of  $Ba_3B_6Si_2O_{16}$  silicoborate were obtained by slow cooling a melt of stoichiometric composition. Crystal structure was determined by direct methods from X-ray single crystal diffractometry data (Bruker KAPPA APEX DUO, MoKa, CCD detector) using a small colorless plate. The structure consists of layered silicoborate anion formed by four crystallographically independent polyhedra: one  $BO_3$  triangle, two  $BO_4$  tetrahedra and one  $SiO_4$  tetrahedron. Ba atoms occupy two different positions in the structure: one of them is inside the layer and another one between the layers. Silicoborate layers are located parallel to (010).



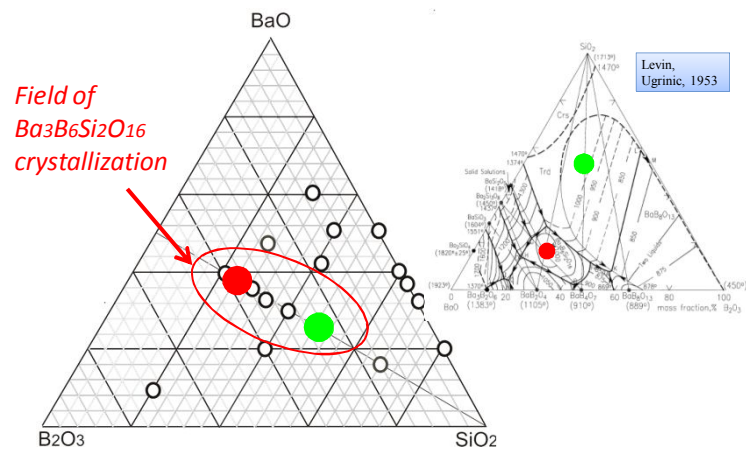
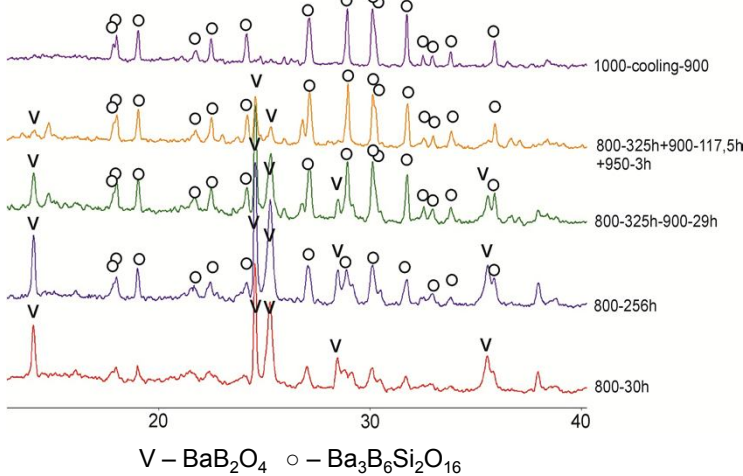
## Crystal Structure

Triclinic,  $P-1$   
 $a = 5.0382(8) \text{ \AA}$   
 $b = 7.6574(12) \text{ \AA}$   
 $c = 8.5262(14) \text{ \AA}$   
 $\alpha = 77.677(5)^\circ$   
 $\beta = 77.879(5)^\circ$   
 $\gamma = 86.324(5)^\circ$   
 $R = 0.07$  (2940 Refl.)  
 $0.09$  (3922 Refl.)

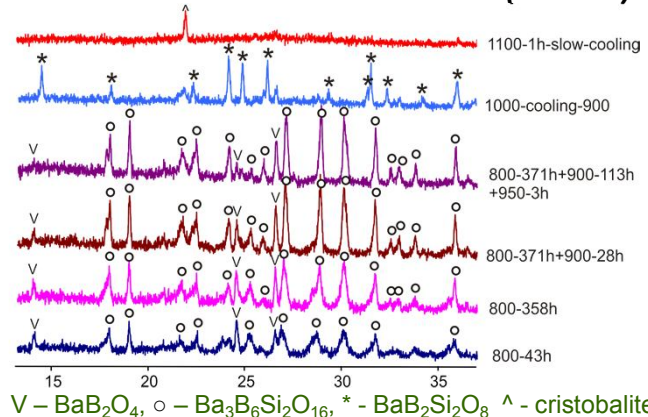
## Phase Formation in $BaO-B_2O_3-SiO_2$ System

According to triangulation reported in (Levin, Ugrinic, 1953)  $Ba_3B_6Si_2O_{16}$  is only one stable compound in the  $BaO-B_2O_3-SiO_2$  ternary system. In a wide field of compositions near  $3BaO \times 3B_2O_3 \times 2SiO_2$  we obtained  $Ba_3B_6Si_2O_{16}$  by solid state reaction at 800, 900 and 950 °C mixed with barium borates and/or silicates depending on the composition of initial mixture. Powder XRD data obtained by us for the pure  $Ba_3B_6Si_2O_{16}$  synthesized by cooling a melt are in well agreement with the reported before in PDF database and with structural data presented here.

### ● $Ba_3B_6Si_2O_{16}$ (3:3:2)



### ● Danburite-like $BaB_2Si_2O_8$ (1:1:2)



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V –  $BaB_2O_4$ , o –  $Ba_3B_6Si_2O_{16}$ , \* –  $BaB_2Si_2O_8$  ^ – cristobalite